



Video Solution on Website:-

<https://physicsaholics.com/home/courseDetails/46>

Video Solution on YouTube:-

<https://youtu.be/u1PnH3mSgPw>

Written Solution on Website:-

<https://physicsaholics.com/note/notesDetails/19>

- Q 1. Mark the correct statement
(a) Rest Mass of photon is zero.
(b) Number of photons is a conserved quantity
(c) Momentum of photon is zero.
(d) None of these
- Q 2. For given energy of photon, $E = 3.03 \times 10^{-19}$ J corresponding wavelength will be: ($h = 6.6 \times 10^{-34}$ sec, $C = 3 \times 10^8$ m/sec.)
(a) 65.6 nm (b) 6.56 nm
(c) 3.4 nm (d) 656 nm
- Q 3. A moving hydrogen atom absorbs a photon of wavelength 122 nm and comes to rest. Then speed of moving hydrogen was
(a) 3.25 m/s (b) 6.5 m/s
(c) 1.75 m/s (d) 8.25 m/s
- Q 4. The number of photons of light having wavelength 100nm which can provide 1J energy is nearly:
(a) 10^7 photons (b) 5×10^{20} photons
(c) 5×10^{17} photons (d) 5×10^7 photons
- Q 5. The equation $E = pc$ is valid -
(a) for an electron as well as for photon
(b) for an electron but not for a photon
(c) for a photon but not for an electron
(d) neither for an electron nor for a photon
- Q 6. The energy of photon of visible light (400 nm to 700 nm) with maximum wavelength in eV is
(a) 1 (b) 1.77 (c) 3.2 (d) 7
- Q 7. What is the momentum of a photon having frequency 1.5×10^{13} Hz?
(a) 3.3×10^{-29} kg m/s (b) 3.3×10^{-34} kg m/s
(c) 6.6×10^{-34} kg m/s (d) 6.6×10^{-30} kg m/s
- Q 8. The energy of photon having $\lambda = 620\text{\AA}$ in eV is
(a) 10
(b) 20
(c) 2



(d) 1

Q 9. A TV station is operated at 100 MW with a signal frequency of 10 Mhz. Calculate the number of photons radiated per second by its antenna?

- (a) 2.5×10^{34}
- (b) 1.5×10^{34}
- (c) 5×10^{34}
- (d) 6×10^{34}

Q 10. Wavelength emitted by a bulb is halved and power is doubled then number of photons emitted (per second) by it will

- (a) Halved
- (b) Doubled
- (c) Quadrupled
- (d) Remain same

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Answer Key

Q.1 a	Q.2 d	Q.3 a	Q.4 c	Q.5 c
Q.6 b	Q.7 a	Q.8 b	Q.9 b	Q.10 d